

Preparation and dielectric characterization of BaLaAlO₄ ceramics

Y. J. Liu · X. M. Chen · Y. Xiao

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Abstract In the present work, BaLaAlO₄ ceramics with orthorhombic structure similar to K₂SO₄ in space group P2₁2₁2₁ were prepared by a solid state sintering process. The dense BaLaAlO₄ ceramics with minor amount of secondary phase have a low dielectric loss and a temperature stable dielectric constant with obvious frequency dependence. A dielectric constant around 15 was obtained at 12 GHz in the present ceramics together with a Qf value over 5,000 GHz.

Keywords BaLaAlO₄ · Microwave dielectric properties · Ceramics · Sintering · Microstructures

1 Introduction

Recently, a group of compounds with general chemical formula ABCO₄ (A=Ca or Sr; B=Y, La, Nd or Sm; C=Al or Ga) and tetragonal K₂NiF₄-type structure within space group I4/mmm have attracted much scientific attention because of their potential applications as substrates for high-T_c superconductor thin films [1–10]. In the author's group, SrRAIO₄ (R = La, Nd, Sm) and CaRAIO₄ (R = Nd, Sm, Y) ceramics have been systematically investigated for microwave resonator application, and the best microwave dielectric characteristics were obtained in SrSmAlO₄ and CaSmAlO₄ among these ceramics [11–13].

On the other hand, compounds with the similar compositions of BaRAIO₄ and BaRGAO₄ (R = rare earth elements)

are not tetragonal with K₂NiF₄ structure, but orthorhombic structure similar to K₂SO₄ in space group P2₁2₁2₁ [8,14]. It is an interesting issue to investigate the dielectric properties of BaRAIO₄ and BaRGAO₄.

In the present work, SrLnAlO₄ (Ln = Nd and Sm) ceramics are prepared by the solid state reaction method, and the microwave dielectric properties are characterized together with the microstructures.

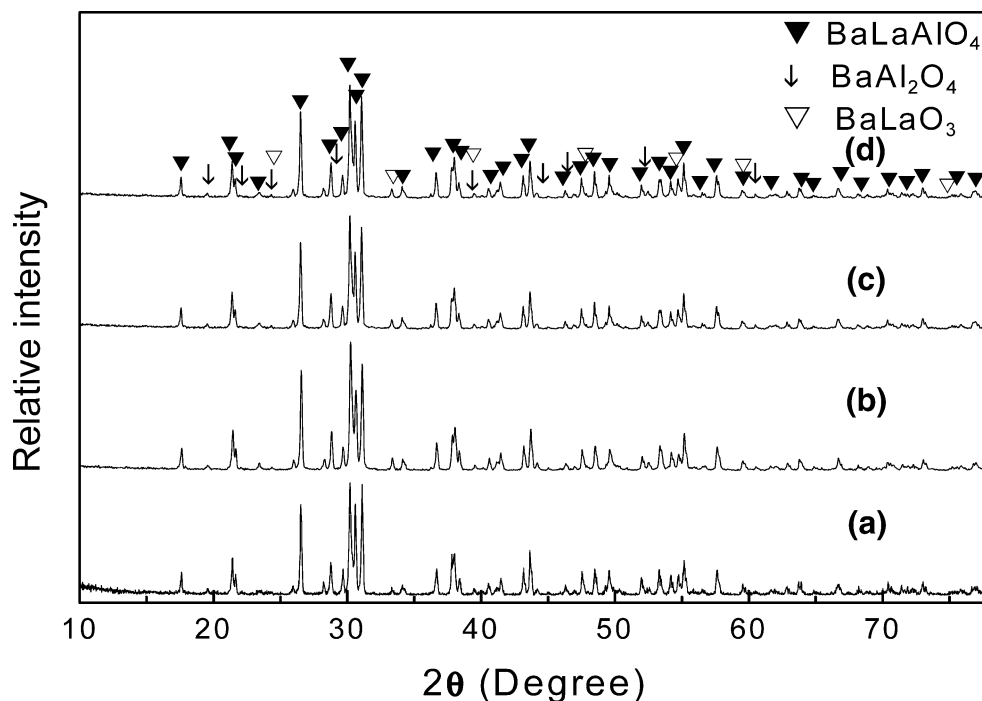
2 Experimental procedure

BaLaAlO₄ ceramics were prepared by a solid-state reaction process using reagent-grade BaCO₃ (99.93%), La₂O₃ (99.99%), and Al₂O₃ (>99.98%) powders as the raw materials. The weighed raw materials were mixed by ball milling with zirconia media in distilled water for 24 h, and the mixtures were heated at 1350 °C in air for 6 h after drying. The calcined powders, with 6 wt% of PVA added, were pressed into disks measuring 12 mm in diameter and 2–6 mm high and then sintered at 1350 °C–1550 °C in air for 3 h. After cooling from the sintering temperature to 1100 °C at a rate of 2 °C/min, the ceramics were cooled inside the furnace.

Ground and thermal-etched surfaces of the sintered samples were observed by scanning electron microscopy (SEM), and the crystal phases were determined by powder x-ray diffraction (XRD) using Cu-Kα radiation after crushing and grinding. Dielectric properties were first measured at 10, 100 kHz and 1 MHz by an LCR meter (HP4284A) in the temperatures range of –50~100 °C, and then the microwave dielectric constant ε and quality factor Q (the inverse of dielectric loss, tanδ) were evaluated around 10 GHz using the resonator method [15]. Because Q factor generally varies inversely with the frequency, in the

Y. J. Liu · X. M. Chen (✉) · Y. Xiao
Department of Materials Science and Engineering,
Zhejiang University,
Hangzhou 310027, China
e-mail: xmchen@cmsce.zju.edu.cn

Fig. 1 XRD patterns of BaLaAlO₄ ceramics: (a) powders calcined at 1350 °C in air for 6 h, (b) sintered at 1475 °C in air for 3 h, (c) sintered at 1500 °C in air for 3 h and (d) sintered at 1525 °C in air for 3 h



microwave region, the product of Qf was used to evaluate the dielectric loss instead of Q.

3 Results and discussion

XRD patterns of BaLaAlO₄ ceramics are shown in Fig. 1. BaLaAlO₄ major phase is observed together with minor amount of BaAl₂O₄ and LaAlO₃ secondary phases. With varying sintering temperature, no significant difference can be distinguished in XRD patterns. The highest bulk density of 5.4718 g/cm³ was obtained in the samples sintered at 1500 °C in air for 3 h.

SEM micrographs on fractured surfaces of BaLaAlO₄ ceramics are shown in Fig. 2, which display the microstructure varying with sintering temperature. The grain growth can be obviously observed. When sintered at 1475 °C, some grains are still cuboids distinct from the ground grains. The grains grow more integrated and bigger as the sintering temperature increases.

Figure 3 gives the frequency dependence of dielectric characteristics of BaLaAlO₄ ceramics sintered at various temperatures. Both dielectric constant and dielectric loss show the frequency dependence, and they tend to be stable as the frequency is above 10 kHz. Sintering temperature has a significant influence on dielectric characteristics. Dense BaLaAlO₄ ceramics show a higher dielectric constant around 18. Sintering temperature shows significant effects on the frequency dependence of dielectric characteristics in BaLaAlO₄ ceramics, the strongest frequency dependence is determined in the BaLaAlO₄ ceramics sintered at

1525 °C, which may be resulted from the over sintering. Moreover, the dielectric constant is temperature stable between −40 and 100 °C, and the temperature dependence of dielectric constant increases with decreasing frequency (see Fig. 4).

The microwave dielectric properties of BaLaAlO₄ ceramics sintered at different temperatures are listed in Table 1. At microwaves, a dielectric constant around 15 is observed, which is much lower than that at 1 MHz and lower frequencies, while the dielectric loss is much higher. The Qf value is 5,000 GHz at 10.5 to 12.5 GHz, and the best one is 5,259 GHz for BaLaAlO₄ ceramics sintered at 1500 °C.

Compared with SrLaAlO₄ ($\epsilon=17.1$, $\tan\delta=0.00035$, $Qf=30770$ GHz), SrSmAlO₄ ($\epsilon=18.8$, $\tan\delta=0.00016$, $Qf=54880$ GHz), SrNdAlO₄ ($\epsilon=17.8$, $\tan\delta=0.00040$, $Qf=25700$ GHz) and CaNdAlO₄ ($\epsilon=18.1$, $\tan\delta=0.00064$, $Qf=16855$ GHz) [11, 13], the Qf and ϵ values of BaLaAlO₄ ceramics are much lower than that for SrAlO₄ (R = La, Sm) and CaNdAlO₄ ceramics [5, 8, 13, 16], while the $\tan\delta$ is much higher. The great distinctness should result from the apparently different structure.

BaLaAlO₄ has a different structure from most Sr- and Ca-based ABCO₄ compounds. According to literatures [5, 8, 13, 15], Sr- and Ca-based ABCO₄ compounds generally have the tetragonal structure with space group I4/mmm. However, BaLaAlO₄ and BaNdGaO₄ crystal are not tetragonal with CaTiO₃ or K₂NiF₄ structure, but orthorhombic similar to K₂SO₄ in space group P2₁2₁2₁ [8, 14]. The coordination polyhedrons of both are octahedron and tetrahedron, respectively.

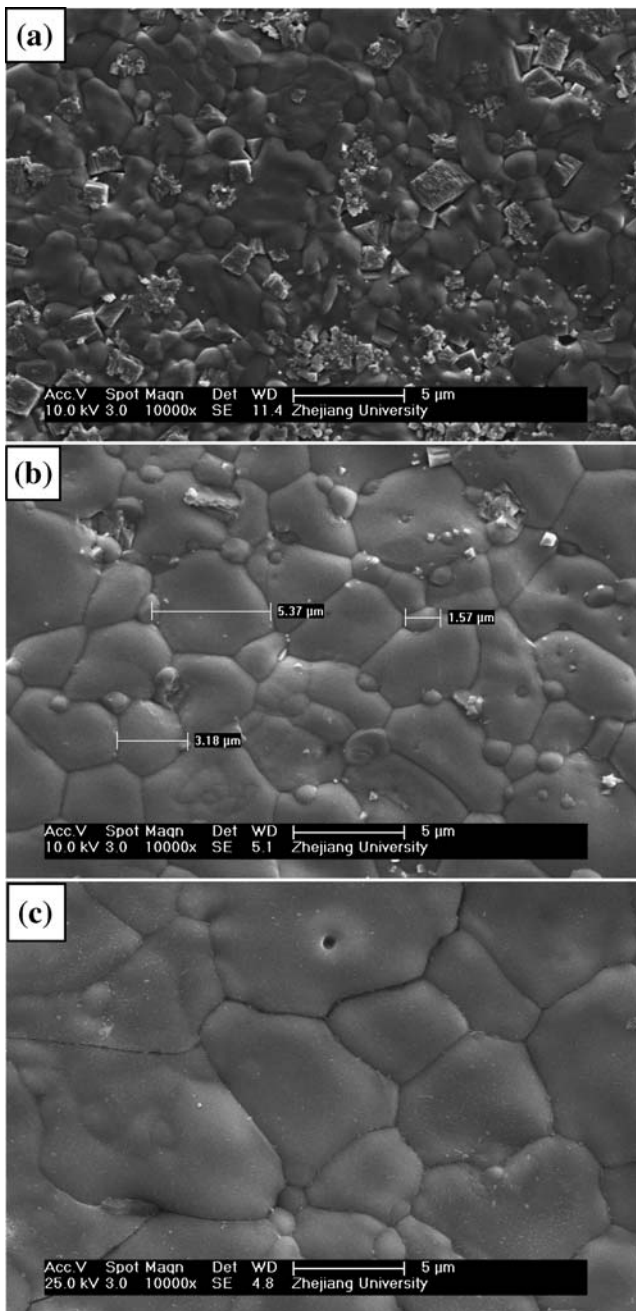


Fig. 2 SEM micrographs of BaLaAlO₄ ceramics sintered at (a) 1475 °C, (b) 1500 °C and (c) 1525 °C in air for 3 h

So in a K₂NiF₄-type ABCO₄ crystal cell, all the CO₆ octahedrons have the same orientation, in which C-atoms tend to shift in the same direction when affected by external electric field, making greater polarization and lower dielectric loss. In contrast, CO₄ tetrahedrons in a K₂SO₄-type ABCO₄ cell have different orientations and small space, thus it is hard for the C-atoms to behave the same in external electric field. So the displacement of C-atoms in K₂SO₄-type is resisted stronger than that in K₂NiF₄ type. As a result, it is hard for K₂SO₄-type ABCO₄ to polarize and therefore the dielectric loss is larger.

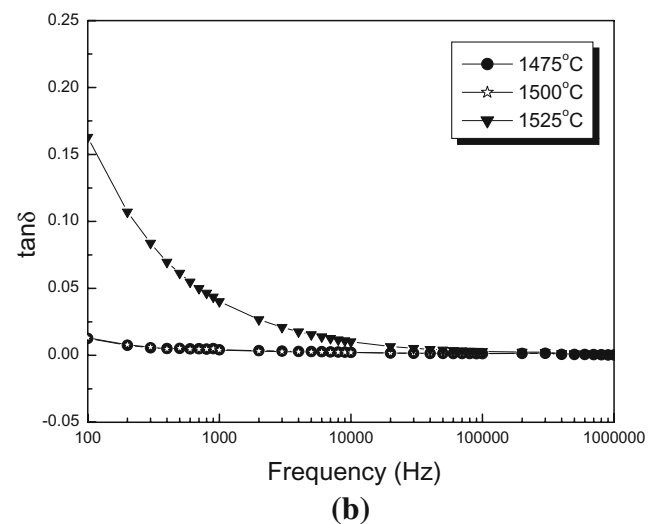
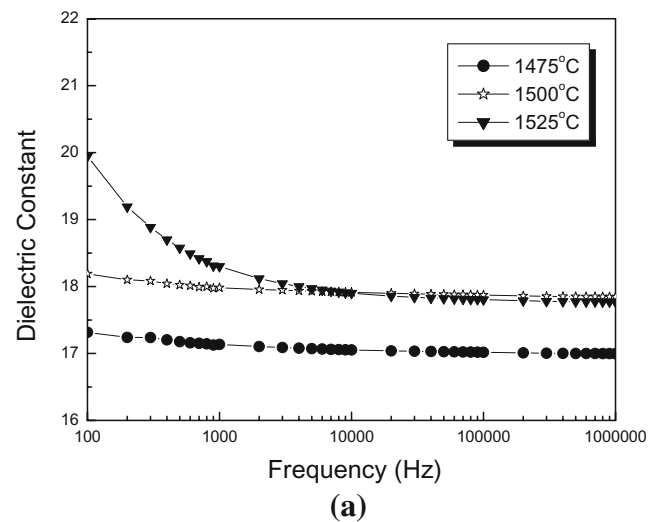


Fig. 3 Frequency dependence of dielectric characteristics of BaLaAlO₄ ceramics sintered at various temperatures: (a) dielectric constant, (b) dielectric loss

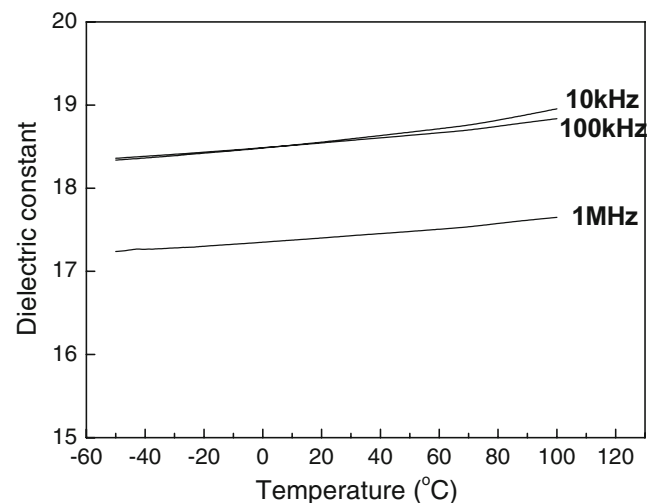


Fig. 4 Temperature dependence of dielectric constant of BaLaAlO₄ ceramics sintered at 1500 °C in air for 3 h

Table 1 Microwave dielectric characteristics of BaLaAlO₄ ceramics sintered on various conditions.

Sintering condition	f_0 (GHz)	E	$\tan\delta$	Qf (GHz)
1475°C/3 h	12.57	14.2	0.0024	5,238
1500°C/3 h	11.57	15.4	0.0022	5,259
1525°C/3 h	10.54	16.5	0.0022	4,977

4 Conclusion

BaLaAlO₄ ceramics were prepared by a solid state sintering process, and the major phase with orthorhombic structure similar to K₂SO₄ in space group P2₁2₁2₁ was observed together with minor amount of BaLaO₃ and BaAl₂O₄. A low dielectric loss was determined combined with a temperature stable dielectric constant with slight frequency dependence in the present ceramics, and dielectric constant around 15 was obtained at 12 GHz together with a Qf value over 5,000 GHz. The improved Qf value is expected through the microstructure optimizing especially eliminating the secondary phases.

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